**S.6 RESOURCEFUL MOCKS CONFIDENTIAL**

**CHEMISTRY PRACTICAL**

**P525/3**

1. BA1 = 3.1g of KMnO4 in 1 litre of water

BA2 = 19.6gl-1 ammonia ferrous sulphate – 6 – water

Thermometers (-10-110oc)

Volumetric flask (250ml)

200cm3 of 2M H2SO4

X=solid oxatic acid (H2C2O42H2))

Weighing scale

1. T = Ni (CH3COO)2: Al2(SO4)3 = 1: 2
2. Q = Benzaldehyde.

Heat Source

**Reagents.**

* Dilute NaoH
* Ammonia solution
* Conc. H2SO4
* Ethanol
* Iron (III) chloride solution
* Litmus solution
* Dilute HCL
* KI
* Barium nitrate situation
* Dimethyl glyoxime
* Fehling solution
* sodium hydrogen sulphite
* Brady’s reagent
* Iodine solution.
* filter paper
* Blue and red litmus papers.

**END**